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HYDROXYCARBONYLPHENYL SUBSTITUTED  
4-(AMINOMETHYL)-PIPERIDINE BENZAMIDES AS 5HT<sub>4</sub>-ANTAGONISTS

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5 The present invention is concerned with novel compounds of formula (I) having 5HT<sub>4</sub>-antagonistic properties. The invention further relates to methods for preparing such novel compounds, pharmaceutical compositions comprising said novel compounds as well as the use as a medicine of said compounds.

10 WO-00/37461 discloses bicyclic benzamides of 3- or 4-substituted 4-(aminomethyl)-piperidine derivatives having 5HT<sub>4</sub>-antagonistic properties.

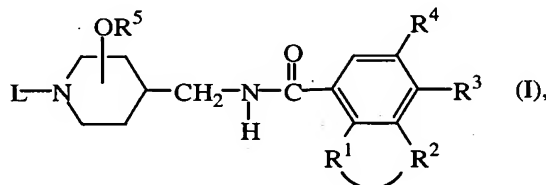
The compounds of the present invention differ structurally from the cited art-known compounds by the presence of a different L radical moiety.

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Unexpectedly, the present compounds of formula (I) have improved metabolic stability properties compared with the compounds disclosed in WO-00/37461.

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The present invention concerns compounds of formula (I)



a stereochemically isomeric form thereof, an *N*-oxide form thereof, or a pharmaceutically acceptable acid or base addition salt thereof,

25

wherein

-R<sup>1</sup>-R<sup>2</sup>- is a bivalent radical of formula

30

- |  |        |
|--|--------|
| -O-CH <sub>2</sub> -O-   | (a-1), |
| -O-CH <sub>2</sub> -CH <sub>2</sub> -  | (a-2), |
| -O-CH <sub>2</sub> -CH <sub>2</sub> -O-  | (a-3), |
| -O-CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -                                   | (a-4), |
| -O-CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -O-                                 | (a-5), |
| -O-CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -                  | (a-6), |
| -O-CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -O-                | (a-7), |
| -O-CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> - | (a-8), |

wherein in said bivalent radicals optionally one or two hydrogen atoms on the same or a different carbon atom may be replaced by C<sub>1-6</sub>alkyl or hydroxy,

R<sup>3</sup> is hydrogen, halo, C<sub>1-6</sub>alkyl or C<sub>1-6</sub>alkyloxy;

R<sup>4</sup> is hydrogen, halo, C<sub>1-6</sub>alkyl; C<sub>1-6</sub>alkyl substituted with cyano, or C<sub>1-6</sub>alkyloxy;

5 C<sub>1-6</sub>alkyloxy; cyano; amino or mono or di(C<sub>1-6</sub>alkyl)amino;

R<sup>5</sup> is hydrogen or C<sub>1-6</sub>alkyl, and the -OR<sup>5</sup> radical is situated at the 3- or 4-position of the piperidine moiety;

L is a radical of formula

	-Alk-R <sup>6</sup>	(b-1),
10	-Alk-X-R <sup>7</sup>	(b-2),
	-Alk-Y-C(=O)-R <sup>9</sup>	(b-3),

wherein each Alk is C<sub>1-12</sub>alkanediyl; and

R<sup>6</sup> is aryl;

R<sup>7</sup> is aryl;

15 X is O, S, SO<sub>2</sub> or NR<sup>8</sup>; said R<sup>8</sup> being hydrogen or C<sub>1-6</sub>alkyl;

R<sup>9</sup> is aryl;

Y is a direct bond, O, S, or NR<sup>10</sup> wherein R<sup>10</sup> is hydrogen or C<sub>1-6</sub>alkyl; and

aryl represents phenyl substituted with 1, 2 or 3 substituents each independently selected from hydroxycarbonyl.

20

As used in the foregoing definitions halo is generic to fluoro, chloro, bromo and iodo;

C<sub>1-4</sub>alkyl defines straight and branched chain saturated hydrocarbon radicals having

from 1 to 4 carbon atoms such as, for example, methyl, ethyl, propyl, butyl, 1-methyl-

ethyl, 2-methylpropyl and the like; C<sub>1-6</sub>alkyl is meant to include C<sub>1-4</sub>alkyl and the

25 higher homologues thereof having 5 or 6 carbon atoms, such as, for example, 2-methyl-

butyl, pentyl, hexyl and the like; C<sub>1-12</sub>alkanediyl defines bivalent straight or branched chain hydrocarbon radicals containing from 1 to 12 carbon atoms such as, for example,

methanediyl, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl, 1,5-pentanediyl,

1,6-hexanediyl, 1,7-heptanediyl, 1,8-octanediyl, 1,9-nonanediyl, 1,10-decanediyl,

30 1,11-undecanediyl, 1,12-dodecanediyl and the branched isomers thereof. C<sub>1-4</sub>alkanediyl

defines bivalent straight or branched chain hydrocarbon radicals containing from 1 to 4 carbon atoms such as, for example, methanediyl, 1,2-ethanediyl, 1,3-propanediyl, and 1,4-butanediyl.

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The term "stereochemically isomeric forms" as used hereinbefore defines all the possible isomeric forms which the compounds of formula (I) may possess. Unless otherwise mentioned or indicated, the chemical designation of compounds denotes the

mixture of all possible stereochemically isomeric forms, said mixtures containing all diastereomers and enantiomers of the basic molecular structure. More in particular, stereogenic centers may have the R- or S-configuration; substituents on bivalent cyclic (partially) saturated radicals may have either the cis- or trans-configuration.

- 5 Compounds encompassing double bonds can have an E or Z-stereochemistry at said double bond. Stereochemically isomeric forms of the compounds of formula (I) are obviously intended to be embraced within the scope of this invention.

10 The pharmaceutically acceptable acid and base addition salts as mentioned hereinabove are meant to comprise the therapeutically active non-toxic acid and base addition salt forms which the compounds of formula (I) are able to form. The pharmaceutically acceptable acid addition salts can conveniently be obtained by treating the base form with such appropriate acid. Appropriate acids comprise, for example, inorganic acids such as hydrohalic acids, e.g. hydrochloric or hydrobromic acid, sulfuric, nitric,  
15 phosphoric and the like acids; or organic acids such as, for example, acetic, propanoic, hydroxyacetic, lactic, pyruvic, oxalic (*i.e.* ethanedioic), malonic, succinic (*i.e.* butanedioic acid), maleic, fumaric, malic, tartaric, citric, methanesulfonic, ethanesulfonic, benzenesulfonic, *p*-toluenesulfonic, cyclamic, salicylic, *p*-aminosalicylic, pamoic and the like acids.

20 Conversely said salt forms can be converted by treatment with an appropriate base into the free base form.

25 The compounds of formula (I) containing an acidic proton may also be converted into their non-toxic metal or amine addition salt forms by treatment with appropriate organic and inorganic bases. Appropriate base salt forms comprise, for example, the ammonium salts, the alkali and earth alkaline metal salts, e.g. the lithium, sodium, potassium, magnesium, calcium salts and the like, salts with organic bases, e.g. the benzathine, *N*-methyl-D-glucamine, hydrabamine salts, and salts with amino acids such  
30 as, for example, arginine, lysine and the like.

The term addition salt as used hereinabove also comprises the solvates which the compounds of formula (I) as well as the salts thereof, are able to form. Such solvates are for example hydrates, alcoholates and the like.

35 Some of the compounds of formula (I) may also exist in their tautomeric form. Such forms although not explicitly indicated in the above formula are intended to be included within the scope of the present invention.

The *N*-oxide forms of the compounds of formula (I), which may be prepared in art-known manners, are meant to comprise those compounds of formula (I) wherein one or several nitrogen atoms are oxidized to the *N*-oxide. Particularly those *N*-oxides are envisaged wherein the piperidine-nitrogen is *N*-oxidized.

A group of interesting compounds consists of those compounds of formula (I) wherein one or more of the following restrictions apply :

- a)  $-R^1-R^2-$  is a radical of formula (a-5); and/or
- 10 b)  $R^3$  is hydrogen, halo, methyl, or methoxy; and/or
- c)  $R^4$  is hydrogen, halo, or methyl; and/or
- d)  $R^4$  is fluoro; and/or
- e) aryl represents phenyl substituted with hydroxycarbonyl; and/or
- f)  $R^5$  is hydrogen or methyl, and the  $-OR^5$  radical is situated at the 3- or 4-position of the piperidine ring; and/or
- 15 g)  $R^5$  is hydrogen or methyl, and the  $-OR^5$  radical is situated at the 3-position of the piperidine ring; and/or
- h) the  $-OR^5$  radical, wherein  $R^5$  is hydrogen or methyl, is situated at the 3-position of the piperidine ring and is in the trans position in relation to the methylene on the 4-position of the piperidine moiety; and/or
- 20 i) the  $-OR^5$  radical, wherein  $R^5$  is hydrogen or methyl, is situated at the 3-position of the piperidine ring and is in the trans position in relation to the methylene on the 4-position of the piperidine moiety and the absolute configuration of said piperidine moiety is (3S, 4S); and/or
- 25 j) L is a radical of formula (b-2) wherein Alk is  $C_{1-4}$ alkanediyl, X represents O, and and  $R^7$  is aryl wherein aryl is phenyl substituted with hydroxycarbonyl.

Other interesting compounds are those compounds of formula (I) wherein  $-R^1-R^2-$  is a bivalent radical of formula



$R^3$  is hydrogen, halo,  $C_{1-6}$ alkyl or  $C_{1-6}$ alkyloxy;

$R^4$  is hydrogen, halo, or  $C_{1-6}$ alkyl;

$R^5$  is hydrogen or  $C_{1-6}$ alkyl, and the  $-OR^5$  radical is situated at the 3- or 4-position of the piperidine moiety;

35 L is a radical of formula



wherein each Alk is  $C_{1-12}$ alkanediyl; and

$R^7$  is aryl;

X is O;

aryl represents phenyl substituted with hydroxycarbonyl.

Particular compounds are those compounds of formula (I) wherein the -OR<sup>5</sup> radical, preferably representing hydroxy or methoxy, is situated at the 3-position of the piperidine moiety having the trans configuration, *i.e.* the -OR<sup>5</sup> radical is in the trans position in relation to the methylene on the piperidine moiety.

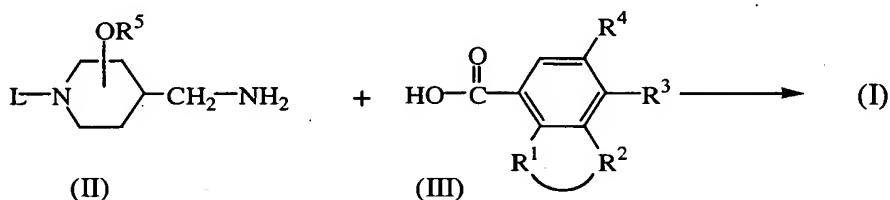
More particular compounds are those compounds of formula (I) wherein the bivalent radical -R<sup>1</sup>-R<sup>2</sup>- is a radical of formula (a-5), the -OR<sup>5</sup> radical represents hydroxy and is situated at the 3-position of the piperidine moiety having the (3S-trans) configuration which corresponds to absolute (3S, 4S) configuration of said piperidine moiety.

Preferred compounds are those more particular compounds wherein L is a radical of formula (b-2) wherein Alk is C<sub>1-4</sub>alkanediyl, and R<sup>7</sup> is aryl wherein aryl is phenyl substituted with hydroxycarbonyl.

More preferred compounds are those preferred compounds wherein Alk is 1,3-propanediyl or 1,4-butanediyl and R<sup>7</sup> is aryl wherein aryl is phenyl substituted with hydroxycarbonyl situated at the 3- or 4-position of the phenyl moiety.

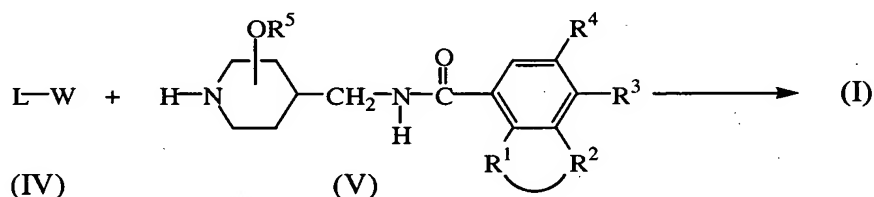
Most preferred compounds are those more preferred compounds wherein Alk is 1,3-propanediyl.

The compounds of formula (I) can be prepared by reacting an intermediate of formula (II) with an carboxylic acid derivative of formula (III) or, optionally a reactive functional derivative thereof, such as, *e.g.* carbonyl imidazole derivatives, acyl halides or mixed anhydrides. Said amide-bond formation may be performed by stirring the reactants in an appropriate solvent, optionally in the presence of a base, such as triethylamine. The hydroxycarbonyl group present in substituents R<sup>6</sup>, R<sup>7</sup> and R<sup>9</sup> is usually protected in the above described reaction sequence in the form of an ester which is removed after the amide-bond formation reaction by hydrolysis under basic conditions.



The hydroxycarbonyl group present in substituents R<sup>6</sup>, R<sup>7</sup> and R<sup>9</sup> is usually protected in the above described reaction sequence in the form of an ester which is removed after the amide-bond formation reaction by hydrolysis under basic conditions.

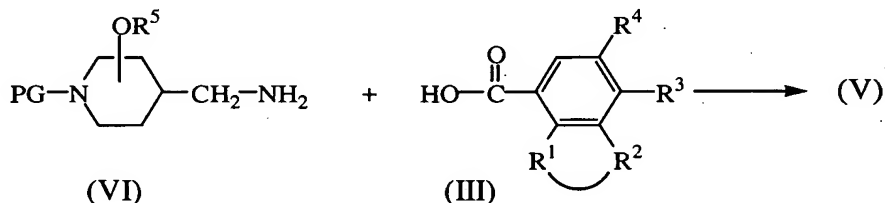
- 5 Also compounds of formula (I) can generally be prepared by *N*-alkylating an intermediate of formula (V) with an intermediate of formula (IV), wherein W is an appropriate leaving group such as, for example, halo, e.g. fluoro, chloro, bromo, iodo, or in some instances W may also be a sulfonyloxy group, e.g. methanesulfonyloxy, benzenesulfonyloxy, trifluoromethanesulfonyloxy and the like reactive leaving groups. The reaction can be performed in a reaction-inert solvent such as, for example, acetonitrile, 2-pentanol, isobutanol, dimethyl acetamide or DMF, and optionally in the presence of a suitable base such as, for example, sodium carbonate, potassium carbonate, N-methylpyrrolidone or triethylamine. Stirring may enhance the rate of the reaction. The reaction may conveniently be carried out at a temperature ranging between room temperature and the reflux temperature of the reaction mixture.



- 20 The hydroxycarbonyl group present in substituents R<sup>6</sup>, R<sup>7</sup> and R<sup>9</sup> is usually protected in the above described reaction sequence in the form of an ester which is removed after the amide-bond formation reaction by hydrolysis under basic conditions.

- 25 Intermediates of formula (V) can be prepared by reacting an intermediate of formula (VII), wherein PG represents an appropriate art-known protective group, such as for example a *tert*-butoxycarbonyl or a benzyl group or a photoremovable group, with an acid of formula (III), or an appropriate reactive functional derivative thereof, such as for example carbonyl imidazole derivatives, and subsequent deprotection of the thus formed intermediate, *i.e.* removal of PG by art-known methods.

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The compounds of formula (I) may further be prepared by converting compounds of formula (I) into each other according to art-known group transformation reactions.

5 The starting materials and some of the intermediates are known compounds and are commercially available or may be prepared according to conventional reaction procedures generally known in the art. For example, intermediates of formula (II) of (VI) can be prepared according to the methodologies described in WO-99/02156 or WO-00/37461.

10 Intermediates of formula (VI) can be prepared according to the general methodology described in WO-99/02156 or WO-00/37461 for the therein described intermediates of formula (VIII).

15 The compounds of formula (I) as prepared in the hereinabove described processes may be synthesized in the form of racemic mixtures of enantiomers which can be separated from one another following art-known resolution procedures. The racemic compounds of formula (I) may be converted into the corresponding diastereomeric salt forms by reaction with a suitable chiral acid. Said diastereomeric salt forms are subsequently separated, for example, by selective or fractional crystallization and the enantiomers are  
20 liberated therefrom by alkali. An alternative manner of separating the enantiomeric forms of the compounds of formula (I) involves liquid chromatography using a chiral stationary phase. Said pure stereochemically isomeric forms may also be derived from the corresponding pure stereochemically isomeric forms of the appropriate starting materials, provided that the reaction occurs stereospecifically. Preferably if a specific  
25 stereoisomer is desired, said compound will be synthesized by stereospecific methods of preparation. These methods will advantageously employ enantiomerically pure starting materials.

30 The compounds of formula (I), the *N*-oxide forms, the pharmaceutically acceptable acid or base addition salts and stereoisomeric forms thereof possess 5HT<sub>4</sub>-antagonistic properties as described in Example C.1.

35 Furthermore the compounds of formula (I) have shown improved metabolic stability as described in Example C.2. These advantageous metabolic stability properties result in a reduced risk of drug-drug interaction on the level of cytochrome P450 enzymes such as e.g. CYP1A2, CYP3A4, CYP2D6, CYP2C9 and CYP2C19 and therefore the present compounds have an improved drug safety profile. Furthermore these advantageous metabolic stability properties may allow for a once daily administration of the

compounds of formula (I) instead of the usual administration of the active ingredient on a regimen of between two or four intakes per day thereby giving more patient compliance.

- 5 In view of the 5HT<sub>4</sub>-antagonistic properties of the compounds of the present invention, the subject compounds may generally be used in the treatment or prophylaxis of gastrointestinal conditions such as hypermotility, irritable bowel syndrome (IBS), constipation- or diarrhea-predominant IBS, pain- and non-pain- predominant IBS, bowel hypersensitivity, and the reduction of pain associated with gastrointestinal  
10 hypersensitivity and/or hyperactivity.

- It is also believed that the compounds of formula (I) are useful in the prevention or prophylaxis of a disturbed, hampered or impaired gastric accommodation such as dyspepsia. Dyspeptic symptoms are for example epigastric pressure, a lack of appetite,  
15 feeling of fullness, early satiety, nausea, vomiting, bloating and gaseous eructation.

The compounds of formula (I) may also be of use in the treatment of other 5HT<sub>4</sub>-related disorders such as boulimia and hyperphagia.

- 20 In view of the utility of the compounds of formula (I), it follows that the present invention also provides a method of treating warm-blooded animals, including humans, (generally called herein patients) suffering from gastrointestinal conditions such as irritable bowel syndrome (IBS). Consequently a method of treatment is provided for relieving patients suffering from conditions such as hypermotility, irritable bowel  
25 syndrome (IBS), constipation- or diarrhea-predominant IBS, pain- and non-pain- predominant IBS, bowel hypersensitivity, and the reduction of pain associated with gastrointestinal hypersensitivity and/or hyperactivity.

- The compounds of formula (I) may also be of potential use in other gastrointestinal  
30 disorders, such as those associated with upper gut motility. In particular, they are of potential use in the treatment of gastric symptoms of gastro-oesophageal reflux disease, such as heartburn (including episodic heartburn, nocturnal heartburn, and meal-induced heartburn).

- 35 Furthermore, the 5HT<sub>4</sub>-antagonistic compounds of formula (I) may also be of potential use in the treatment or prophylaxis of bladder hypersensitivity, overactive bladder, lower urinary tract symptoms, benign prostatic hypertrophy (BPH), prostatitis, detrusor hyperreflexia, outlet obstruction, urinary frequency, nocturia, urinary urgency, pelvic

hypersensitivity, urge incontinence, urethritis, prostatodynia, cystitis, idiopathic bladder hypersensitivity, urinary incontinence or urinary incontinence associated with irritable bowel syndrome. In this respect, it may be advantageous to combine the 5HT<sub>4</sub>-antagonistic compounds of formula (I) with an alpha-adrenoceptor antagonist such as alfuzosin, indoramin, tamsulosin, doxazosin, terazosin, abanoquil, or prazosin in order to obtain pharmaceutical compositions comprising such an alpha-adrenoceptor antagonist, and a 5-HT<sub>4</sub>-receptor antagonist of formula (I).

Hence, the present invention provides compounds of formula (I) for use as a medicine, and in particular the use of compounds of formula (I) for the manufacture of a medicine for treating gastrointestinal conditions such as hypermotility, IBS, constipation- or diarrhea-predominant IBS, pain- and non-pain predominant IBS, bowel hypersensitivity, and the reduction of pain associated with gastrointestinal hypersensitivity and/or hyperactivity. Both prophylactic and therapeutic treatment are envisaged.

In view of the 5HT<sub>4</sub>-antagonistic properties of the compounds of the present invention, the subject compounds may also be of use in treating or preventing 5HT<sub>4</sub>-related CNS disorders in a human. In particular, the compounds of formula (I) can be used to treat a variety of CNS disorders including but not limited to drug substance abuse, cognitive disorders such as Alzheimer's disease, senile dementia; behavioral disorders such as schizophrenia, mania, obsessive-compulsive disorder and psychoactive substance use disorders; mood disorders such as depression, bipolar affective disorder, anxiety and panic disorder; disorders of control of autonomic function such as hypertension and sleep disorders; obsessive/compulsive disorders including anorexia and bulimia, and neuropsychiatric disorders, such as Gilles de la Tourette's syndrome, and Huntington's disease.

To prepare the pharmaceutical compositions of this invention, an effective amount of the particular compound, in base or acid addition salt form, as the active ingredient is combined in intimate admixture with a pharmaceutically acceptable carrier, which carrier may take a wide variety of forms depending on the form of preparation desired for administration. These pharmaceutical compositions are desirably in unitary dosage form suitable, preferably, for administration orally, rectally or by parenteral injection. For example, in preparing the compositions in oral dosage form, any of the usual pharmaceutical media may be employed, such as, for example, water, glycols, oils, alcohols and the like in the case of oral liquid preparations such as suspensions, syrups, elixirs and solutions; or solid carriers such as starches, sugars, kaolin, lubricants, binders, disintegrating agents and the like in the case of powders, pills, capsules and

tablets. Because of their ease in administration, tablets and capsules represent the most advantageous oral dosage unit form, in which case solid pharmaceutical carriers are obviously employed. For parenteral compositions, the carrier will usually comprise sterile water, at least in large part, though other ingredients, for example, to aid  
5 solubility, may be included. Injectable solutions, for example, may be prepared in which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable suspensions may also be prepared in which case appropriate liquid carriers, suspending agents and the like may be employed. In the compositions suitable for percutaneous administration, the carrier optionally comprises a penetration  
10 enhancing agent and/or a suitable wetting agent, optionally combined with suitable additives of any nature in minor proportions, which additives do not cause a significant deleterious effect to the skin. Said additives may facilitate the administration to the skin and/or may be helpful for preparing the desired compositions. These compositions may be administered in various ways, e.g., as a transdermal patch, as a spot-on, as an  
15 ointment. Acid addition salts of (I) due to their increased water solubility over the corresponding base form, are obviously more suitable in the preparation of aqueous compositions.

It is especially advantageous to formulate the aforementioned pharmaceutical  
20 compositions in dosage unit form for ease of administration and uniformity of dosage. Dosage unit form as used in the specification and claims herein refers to physically discrete units suitable as unitary dosages, each unit containing a predetermined quantity of active ingredient calculated to produce the desired therapeutic effect in association with the required pharmaceutical carrier. Examples of such dosage unit forms are  
25 tablets (including scored or coated tablets), capsules, pills, powder packets, wafers, injectable solutions or suspensions, teaspoonfuls, tablespoonfuls and the like, and segregated multiples thereof.

For oral administration, the pharmaceutical compositions may take the form of solid  
30 dose forms, for example, tablets (both swallowable-only and chewable forms), capsules or gelcaps, prepared by conventional means with pharmaceutically acceptable excipients such as binding agents (e.g. pregelatinised maize starch, polyvinylpyrrolidone or hydroxypropyl methylcellulose); fillers (e.g. lactose, microcrystalline cellulose or calcium phosphate); lubricants e.g. magnesium stearate,  
35 talc or silica); disintegrants (e.g. potato starch or sodium starch glycollate); or wetting agents (e.g. sodium lauryl sulphate). The tablets may be coated by methods well known in the art.

Liquid preparations for oral administration may take the form of, for example, solutions, syrups or suspensions, or they may be presented as a dry product for constitution with water or other suitable vehicle before use. Such liquid preparations may be prepared by conventional means, optionally with pharmaceutically acceptable additives such as suspending agents (e.g. sorbitol syrup, methylcellulose, hydroxypropyl methylcellulose or hydrogenated edible fats); emulsifying agents (e.g. lecithin or acacia); non-aqueous vehicles (e.g. almond oil, oily esters or ethyl alcohol); and preservatives (e.g. methyl or propyl p-hydroxybenzoates or sorbic acid).

- 5 Pharmaceutically acceptable sweeteners comprise preferably at least one intense sweetener such as saccharin, sodium or calcium saccharin, aspartame, acesulfame potassium, sodium cyclamate, alitame, a dihydrochalcone sweetener, monellin, stevioside or sucralose (4,1',6'-trichloro-4,1',6'-trideoxygalactosucrose), preferably saccharin, sodium or calcium saccharin, and optionally a bulk sweetener such as sorbitol, mannitol, fructose, sucrose, maltose, isomalt, glucose, hydrogenated glucose syrup, xylitol, caramel or honey.

- Intense sweeteners are conveniently employed in low concentrations. For example, in the case of sodium saccharin, the concentration may range from 0.04% to 0.1% (w/v) based on the total volume of the final formulation, and preferably is about 0.06% in the low-dosage formulations and about 0.08% in the high-dosage ones. The bulk sweetener can effectively be used in larger quantities ranging from about 10% to about 35%, preferably from about 10% to 15% (w/v).

- 25 The pharmaceutically acceptable flavours which can mask the bitter tasting ingredients in the low-dosage formulations are preferably fruit flavours such as cherry, raspberry, black currant or strawberry flavour. A combination of two flavours may yield very good results. In the high-dosage formulations stronger flavours may be required such as Caramel Chocolate flavour, Mint Cool flavour, Fantasy flavour and the like pharmaceutically acceptable strong flavours. Each flavour may be present in the final composition in a concentration ranging from 0.05% to 1% (w/v). Combinations of said strong flavours are advantageously used. Preferably a flavour is used that does not undergo any change or loss of taste and colour under the acidic conditions of the formulation.

- 35 The formulations of the present invention may optionally include an anti-flatulent, such as simethicone, alpha-D-galactosidase and the like.

The compounds of the invention may also be formulated as depot preparations. Such long acting formulations may be administered by implantation (for example subcutaneously or intramuscularly) or by intramuscular injection. Thus, for example, the compounds may be formulated with suitable polymeric or hydrophobic materials (for example as an emulsion in an acceptable oil) or ion exchange resins, or as sparingly soluble derivatives, for example as a sparingly soluble salt.

The compounds of the invention may be formulated for parenteral administration by injection, conveniently intravenous, intramuscular or subcutaneous injection, for example by bolus injection or continuous intravenous infusion. Formulations for injection may be presented in unit dosage form e.g. in ampoules or in multidose containers, with an added preservative. The compositions may take such forms as suspensions, solutions or emulsions in oily or aqueous vehicles, and may contain formulatory agents such as isotonicizing, suspending, stabilising and/or dispersing agents. Alternatively, the active ingredient may be in powder form for constitution with a suitable vehicle, e.g. sterile pyrogen-free water before use.

The compounds of the invention may also be formulated in rectal compositions such as suppositories or retention enemas, e.g. containing conventional suppository bases such as cocoa butter or other glycerides.

For intranasal administration the compounds of the invention may be used, for example, as a liquid spray, as a powder or in the form of drops.

In general it is contemplated that a therapeutically effective amount would be from about 0.0001 mg/kg to about 1 mg/kg body weight, preferably from about 0.001 mg/kg to about 0.5 mg/kg body weight.

#### Experimental part

In the procedures described hereinafter the following abbreviations were used : "ACN" stands for acetonitrile; "THF", which stands for tetrahydrofuran; "DCM" stands for dichloromethane; "DIPE" stands for diisopropylether, "DMF" stands for dimethylformamide, and "DMA" stands for dimethylacetamide.

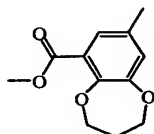
For some chemicals the chemical formula was used, e.g. NaOH for sodium hydroxide, Na<sub>2</sub>CO<sub>3</sub> for sodium carbonate, K<sub>2</sub>CO<sub>3</sub> for potassium carbonate, CuO for cupper(II)oxide, NaNO<sub>2</sub> for sodium nitrite, CH<sub>2</sub>Cl<sub>2</sub> for dichloromethane, CH<sub>3</sub>OH for methanol, NH<sub>3</sub> for ammonia, HCl for hydrochloric acid, NaBF<sub>4</sub> for sodium tetrafluoroborate.

Chiralcel AD is a chiral stationary phase column material purchased from Daicel Chemical Industries, LTd, in Japan.

#### A. Preparation of the intermediates

##### 5 Example A.1

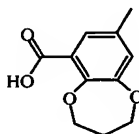
a) Preparation of



intermediate (1)

10 A mixture of methyl 2,3-dihydroxy-5-methylbenzoate (0.198 mol), 1,3-dibromopropane (0.198 mol) and  $K_2CO_3$  (0.396 mol) in 2-propanone (360 ml) was stirred and refluxed for 6 hours, then cooled and the solvent was evaporated. The mixture was poured out into ice water and filtered. The filtrate was extracted with ethyl acetate. The organic layer was separated, dried, filtered, the solvent was evaporated and purified by column chromatography over silica gel (eluent : cyclohexane/ethyl acetate 80/20 to 70/30), yielding intermediate (1).

b) Preparation of

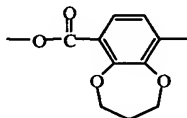


intermediate (2)

15 A mixture of intermediate (1) (0.1129 mol) in a mixture of a NaOH solution 2N (370 ml) and THF (370 ml) was stirred at room temperature for 15 hours. THF was evaporated and the mixture was acidified with HCl 12N. The precipitate was filtered, washed with water and dried, yielding 21.9g of intermediate (2) (mp. 74°C).

##### Example A.2

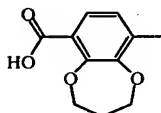
a) Preparation of



intermediate (3)

20 A mixture of 2,3-dihydroxy-4-methyl-benzoic acid methylester (1.2 mol), 1,3-dibromopropane (152 ml) and  $K_2CO_3$  (380 g) in 2-propanone (2500 ml) was stirred and refluxed for 20 hours. The reaction mixture was cooled, filtered and the filtrate was evaporated, yielding 300 g of intermediate (3).

b) Preparation of



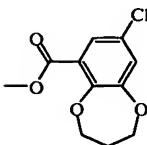
intermediate (4)

25 A mixture of intermediate (3) (1.12 mol) in NaOH (2 M) (1800 ml) and THF (500 ml) was stirred and refluxed for 3 hours. The reaction mixture was cooled and the organic

solvent was evaporated. The aqueous concentrate was acidified with HCl and the resulting precipitate was filtered off, washed with water, and dried, yielding 403 g of intermediate (4).

### 5 Example A.3

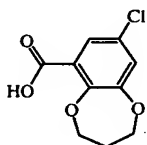
a) Preparation of



intermediate (5)

A mixture of 5-chloro-2,3-dihydroxy-benzoic acid methyl ester (0.3 mol), 1,3-dibromopropane (0.42 mol) and  $K_2CO_3$  (0.66 mol) in 2-propanone (500 ml) was stirred and refluxed for 20 hours, then filtered hot and the filtrate was evaporated. The residue was purified by column chromatography over silica gel (eluent : DCM). The desired fractions were collected and the solvent was evaporated. Toluene was added and azeotroped on the rotary evaporator, yielding 69 g of methyl 8-chloro-3,4-dihydro-2H-1,5-benzodioxepin-6-carboxylate (intermediate 5).

b) Preparation of

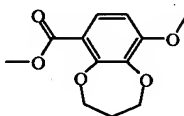


intermediate (6)

A mixture of intermediate (5) (0.25 mol) and KOH (1 mol) in water (650 ml) was stirred and refluxed for 2 hours. The reaction mixture was cooled, acidified with HCl and the resulting precipitate was filtered off, washed with water, and dried, yielding 48 g of 8-chloro-3,4-dihydro-2H-1,5-benzodioxepin-6-carboxylic acid (intermediate 6).

### Example A.4

a) Preparation of

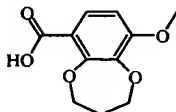


intermediate (7)

A mixture of 2,3-dihydroxy-4-methoxy benzoic acid methyl ester (0.45 mol), 1,3-dibromopropane (0.72 mol),  $K_2CO_3$  (155 g) and CuO (3.6 g) in DMF (2500 ml) was stirred at 120°C to 130°C for 7 hours, cooled and filtered. The solvent was evaporated. HCl (aqueous solution of 0.5 N, 1000 ml) was added. The mixture was extracted twice with DCM (750 ml). The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent : hexane/ethyl acetate/DCM 70/30/15). The pure fractions were collected and the solvent was evaporated. The residue was crystallized from DIPE,

yielding methyl 3,4-dihydro-9-methoxy-2*H*-1,5-benzodioxepin-6-carboxylate (intermediate 7).

b) Preparation of

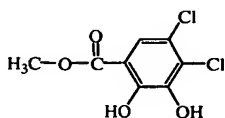


intermediate (8)

A NaOH solution (500 ml, 2N) was added to a solution of intermediate (7) in THF (250 ml). The mixture was stirred at room temperature overnight. The solvent was evaporated partially. The residue was extracted with DCM. The mixture was separated into its layers. The aqueous layer was acidified with a concentrated HCl solution until pH = 1 to 2. The solid was filtered off, washed with water and dried, yielding 35.5 g of 9-methoxy-3,4-dihydro-2*H*-1,5-benzodioxepin-6-carboxylic acid (intermediate 8).

#### Example A.5

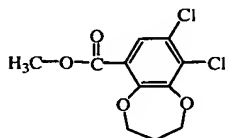
a) Preparation of



intermediate (9)

A mixture of 5-chloro-2,3-dihydroxy benzoic acid methyl ester (0.49 mol), in acetic acid (2000 ml) was stirred and refluxed. A solution of *N*-chlorosuccinimide (0.49 mol) in acetic acid (600 ml) was added dropwise at reflux. The reaction mixture was stirred and refluxed for 30 minutes. Extra solution of *N*-chlorosuccinimide (0.075 mol) in acetic acid (100 ml) was added dropwise at reflux. The reaction mixture was stirred and refluxed for 30 minutes, then cooled and poured out into water (500 ml). The residue was extracted with toluene (3 times). The separated organic layer was washed with water, dried, and evaporated. The residue was crystallized from DIPE and petroleum ether, yielding 70 g of intermediate (9).

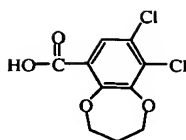
b) Preparation of



intermediate (10)

A mixture of intermediate (9) (0.3 mol), 1,3-dibromopropane (0.35 mol) and K<sub>2</sub>CO<sub>3</sub> (0.7 mol) in 2-propanone (1000 ml) was stirred and refluxed for 30 hours. The reaction mixture was cooled, diluted with water (2000 ml) and extracted twice with DCM. The separated organic layer was washed with water, dried, and the solvent was evaporated. The residue was crystallized from DIPE and petroleum benzene, yielding 55 g of intermediate (10).

c) Preparation of



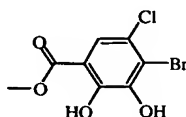
intermediate (11)

A mixture of intermediate (10) (0.2 mol) and KOH (1 mol) in water (1000 ml) was stirred and refluxed for 90 minutes. The reaction mixture was cooled, acidified with HCl and the resulting precipitate was filtered off, washed with water, and dried, yielding 46 g of intermediate (11).

5

Example A.6

a) Preparation of

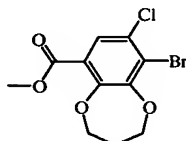


intermediate (12)

A mixture of 5-chloro-2,3-dihydroxy benzoic acid methyl ester (0.1 mol) in acetic acid (250 ml) and *N*-bromosuccinimide (0.11 mol) was stirred and refluxed for 4 hours. The reaction mixture was cooled and poured out into water (500 ml). The precipitate was filtered and dried, yielding 23 g of intermediate (12).

10

b) Preparation of

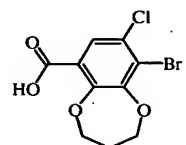


intermediate (13)

A mixture of intermediate (12) (0.7 mol), 1,3-dibromopropane (0.94 mol) and K<sub>2</sub>CO<sub>3</sub> (1.55 mol) in 2-propanone (1300 ml) was stirred and refluxed for 20 hours. The reaction mixture was cooled, filtered and the solvent was evaporated. The residue was solidified in petroleum ether, filtered and dried, yielding 240 g of intermediate (13).

15

c) Preparation of



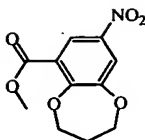
intermediate (14)

A mixture of intermediate (13) (0.053 mol) and KOH (0.2 mol) in water (160 ml) was stirred and refluxed for 90 minutes. The reaction mixture was cooled and the aqueous layer was extracted with DCM. The aqueous layer was acidified with HCl and the resulting precipitate was filtered off, washed with water, and dried, yielding 13 g of intermediate (14).

20

Example A.7

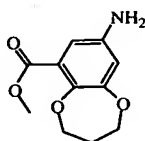
a) Preparation of



intermediate (15)

A mixture of 5-nitro-2,3-dihydroxybenzoic acid methylester (0.3 mol),  $K_2CO_3$  (0.66 mol), 1,3-dibromopropane (0.42 mol) and tetra-n-butylammonium bromide (4.5 g) in 2-propanone (900 ml) and DMA (600 ml) was stirred and refluxed for 30 hours. The reaction mixture was stirred for two days at room temperature and then filtered. The solvent was evaporated and the residue was partitioned between water and DCM. The separated organic layer was dried, filtered and concentrated. The residue was suspended in DIPE, filtered, dried and purified by column chromatography over silica gel (eluent :  $CH_2Cl_2/CH_3OH$  98/2), yielding 33.5g of intermediate (15).

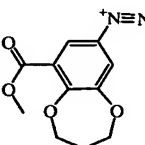
b) Preparation of



intermediate (16)

A mixture of intermediate (15) (0.11 mol) in THF (250 ml) was hydrogenated with palladium-on-carbon 10% (3 g) as a catalyst in the presence of a thiophene-solution (1ml). After uptake of hydrogen (3 equivalents), the catalyst was filtered off over dicalite and the filtrate was concentrated, yielding 24.7g of intermediate (16).

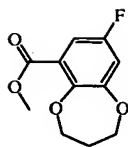
c) Preparation of



intermediate (17)

Intermediate (16) (0.0448 mol) was added portionwise at 5°C to a mixture of concentrated HCl (10 ml) in water (10 ml). The mixture was brought to 0°C. A solution of  $NaNO_2$  (0.048 mol) in water (10 ml) was added dropwise at 0°C. The mixture was stirred at a temperature between 0°C and 5°C for 1 hour, then filtered. The filtrate was cooled to 0°C, then added to a solution of  $NaBF_4$  (0.076 mol) in water (20 ml). The mixture was stirred at 0°C for 30 minutes. The precipitate was filtered, washed with a minimum of water, then with diethyl ether/water (50/50), then with diethyl ether and dried at room temperature under vacuo, yielding 12.10 g of intermediate (17).

d) Preparation of

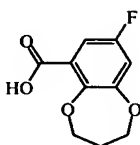


intermediate (18)

A mixture of intermediate (17) (0.0387 mol) and sodium fluoride (0.1549 mol) in toluene (120 ml) was stirred and refluxed overnight, then brought to room temperature. The precipitate was filtered. The filtrate was washed with toluene and evaporated till dryness. The residue was taken up in DCM. The solvent was evaporated till dryness.

- 5 The residue was purified by column chromatography over silica gel (eluent : DCM), yielding 2.8 g of intermediate (18).

e) Preparation of

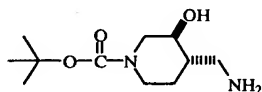


intermediate (19)

- 10 A mixture of intermediate (18) (0.0124 mol) in a NaOH solution (2N, 25 ml) and THF (25 ml) was stirred at room temperature overnight. THF was evaporated and ethyl acetate was added. The mixture was extracted with ethyl acetate, then acidified with HCl till pH 2 was obtained. The precipitate was filtered, washed with water, then with diethyl ether and dried, yielding 2.16 g of intermediate (19).

### Example A.8

Preparation of

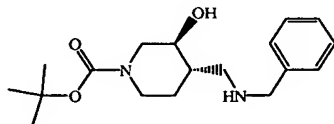


intermediate (20)

- 15 A mixture of 1,1-dimethylethyl (trans)-3-hydroxy-4-[[[(phenylmethyl)amino]methyl]-1-piperidinecarboxylate [described in WO-00/37461 as intermediate (1-d)] (0.023 mol) in methanol (100 ml) was hydrogenated with palladium-on-carbon (10%, 1 g) as a catalyst. After uptake of hydrogen (1 equivalent), the catalyst was filtered off and the filtrate was evaporated. The residue was solidified in DIPE + ACN, filtered off and
- 20 dried, yielding 4 g of 1,1-dimethylethyl (trans)-4-(aminomethyl)-3-hydroxy-1-piperidinecarboxylate (intermediate 20, mp. 178°C).

### Example A.9

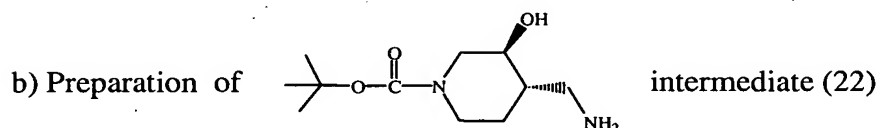
a) Preparation of



intermediate (21)

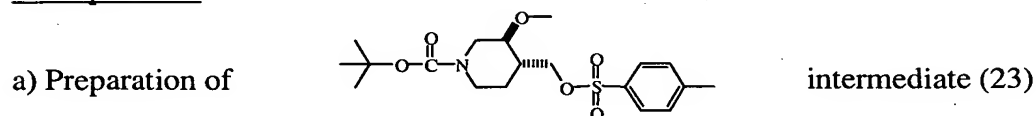
- 25 1,1-Dimethylethyl (trans)-3-hydroxy-4-[[[(phenylmethyl)amino]methyl]-1-piperidinecarboxylate [described in WO-00/37461 as intermediate (1-d)] (2.73 mol)

was separated and purified by chiral column chromatography over Chiralcel AD (eluent : hexane/ethanol 80/20). The desired fractions were collected and the solvent was evaporated. Toluene was added and azeotroped on the rotary evaporator, yielding 377 g of 1,1-dimethylethyl (3S-trans)-3-hydroxy-4-[[[(phenylmethyl)amino]methyl]-1-piperidinecarboxylate (intermediate 21).

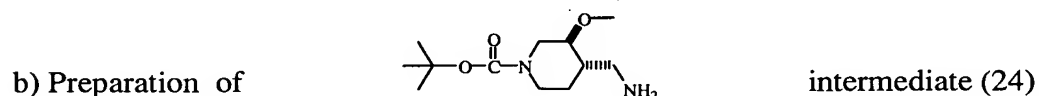


A mixture of intermediate (21) (0.028 mol) in methanol (100 ml) was hydrogenated with palladium-on-carbon (10%, 2 g) as a catalyst. After uptake of hydrogen (1 equivalent) the catalyst was filtered off and the filtrate was evaporated, yielding 4.7 g of 1,1-dimethylethyl (3S-trans)-4-(aminomethyl)-3-hydroxy-1-piperidinecarboxylate (intermediate (22);  $[\alpha]_D^{20} = +4.37^\circ$  (c = 24.03 mg/5 ml in CH<sub>3</sub>OH)).

#### Example A.10



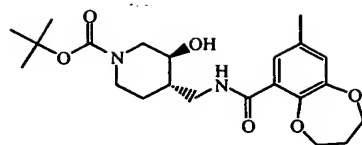
Reaction under nitrogen atmosphere. Sodiumhydride (0.3 mol) was added to a solution of 1,1-dimethylethyl trans-3-hydroxy-4-[[[(4-methylphenyl)sulfonyl]oxy]methyl]-1-piperidinecarboxylate [described in WO-00/37461 as intermediate (1-c)] (0.27 mol) in THF (1300 ml). The mixture was stirred for 30 minutes. Methyl iodide (0.54 mol) was added and the resulting reaction mixture was stirred for 90 minutes. A small amount of water was added. The solvent was evaporated and the residue was partitioned between water and DCM. The organic layer was separated, dried, filtered and the solvent was evaporated, yielding 1,1-dimethylethyl trans-4-[[[(4-methylphenyl)sulfonyl]oxy]methyl]-3-methoxy-1-piperidinecarboxylate (intermediate 23).



A mixture of intermediate (23) (0.065 mol) in THF (250 ml) was treated with liquid NH<sub>3</sub> in an autoclave at 125°C during 16 hours. The reaction mixture was filtered and the filtrate was evaporated. The residue was partitioned between a 5% aqueous NaOH solution and DCM. The organic layer was separated, dried, filtered and the solvent was evaporated, yielding 16 g of 1,1-dimethylethyl (trans)-4-(aminomethyl)-3-methoxy-1-piperidinecarboxylate (intermediate (24)).

Example A.11

a) Preparation of

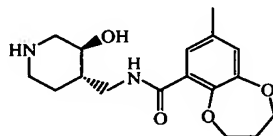


(intermediate 25)

(3S-trans)

5 A mixture of intermediate (2) (0.336 mol) and triethylamine (0.4 mol) in DCM (1000 ml) was stirred at 5°C, then ethyl chloroformate (0.35 mol) was added dropwise and the reaction mixture was stirred for 30 minutes. To this mixture, a solution of intermediate (22) (83 g) in DCM (1000 ml) was added at 5°C, then the reaction mixture was allowed to reach room temperature and was washed with water. The organic layer was separated, dried, filtered and the solvent was evaporated, yielding 150 g of intermediate (25).

b) Preparation of



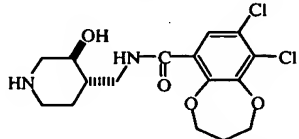
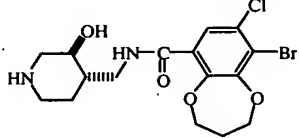
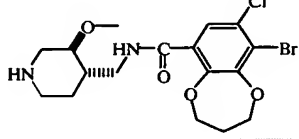
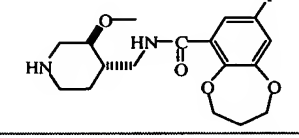
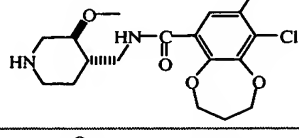
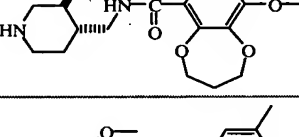
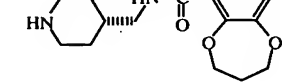
(intermediate 26)

(3S-trans)

15 A mixture of intermediate (25) (0.336 mol) in 2-propanol saturated with HCl (160 ml) and 2-propanol (1400 ml) was stirred and refluxed for 1 hour. The solvent was evaporated and the residue was taken up in a mixture of DCM and a small amount of methanol. The mixture was washed with an aqueous ammonia solution and the organic layer was separated, dried, filtered. The solvent was evaporated, yielding 71 g of intermediate (26).

20 Table I-1 : intermediates (27) to (36) were prepared according to the same procedure of Example A.11

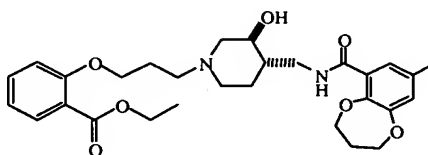
Intmediate	Structure	Physical data
27		trans;
28		3S-trans; mp. 215°C; [α] <sub>D</sub> <sup>20</sup> = -14.03° (c = 23.88 mg/5 ml in methanol)
29		3S-trans; mp. 114°C; [α] <sub>D</sub> <sup>20</sup> = -14.34° (c = 24.41 mg/5 ml in methanol)

Intermediate	Structure	Physical data
30		trans; .HCl (1:1); mp. 173°C
31		trans; .HCl (1:1); mp. 206°C
32		trans;
33		trans;
34		trans; .HCl (1:1); mp. 220°C
35		trans;
36		trans; ; mp. 145°C

## B. Preparation of the final compounds

### Example B.1

a) Preparation of

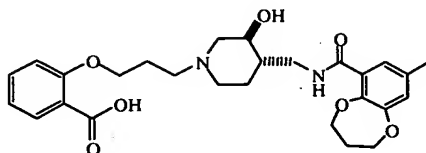


intermediate (37)

- 5 A mixture of intermediate (27) (0.0156 mol), ethyl 2-(3-chloropropoxy)-benzoic acid ester (0.0187 mol) and  $K_2CO_3$  (0.037 mol) in acetonitril (50 ml) was refluxed overnight, poured out into ice water and extracted with ethyl acetate. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent :  $CH_2Cl_2/CH_3OH/NH_4OH$  94/6/0.5).

The pure fractions were collected and the solvent evaporated, yielding intermediate (37) (mp. 116°C).

b) Preparation of

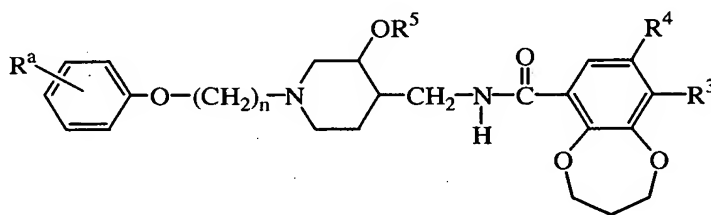


compound (1)

Lithium hydroxide monohydrate (0.0113 mol) was added at room temperature to a mixture of intermediate (38) (0.0056 mol) in THF (30 ml) and water (30 ml). The mixture was stirred at room temperature overnight. THF was evaporated. The mixture was acidified with HCl (3N) and extracted with DCM. The organic layer was separated, dried, filtered, and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent : CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>OH/NH<sub>4</sub>OH 80/20/1 then eluent : CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>OH/NH<sub>4</sub>OH 70/30/2), yielding compound (1) (mp. 114°C).

Table F-1 lists the compounds that were prepared according to the procedure of Example B.1 by reacting one of the intermediates (26) to (36) with one of the following compounds methyl 4-(4-chlorobutoxy)-benzoic acid ester, methyl 4-(4-chloroethoxy)-benzoic acid ester, ethyl 3(3-chloropropoxy)-benzoic acid ester, ethyl 4-(3-chloropropoxy)-benzoic acid ester, or ethyl 2-(3-chloropropoxy)-benzoic acid ester.

Table F-1



20

Co. No.	R <sup>a</sup>	n	R <sup>3</sup>	R <sup>4</sup>	R <sup>5</sup>	Physical data
1	2-HOOC	3	H	CH <sub>3</sub>	H	trans; mp. 114°C
2	4-HOOC	4	H	CH <sub>3</sub>	H	3S-trans; mp. 188°C; [α] <sub>D</sub> <sup>20</sup> = -11.00° (c = 23.63 mg/5 ml in methanol)
3	4-HOOC	4	CH <sub>3</sub>	H	H	3S-trans; mp. 170°C; [α] <sub>D</sub> <sup>20</sup> = -11.71° (c = 23.06 mg/5 ml in methanol)
4	4-HOOC	2	H	Cl	H	3S-trans; mp. 220-225°C

Co. No.	R <sup>a</sup>	n	R <sup>3</sup>	R <sup>4</sup>	R <sup>5</sup>	Physical data
5	4-HOOC	4	H	Cl	H	3S-trans; mp. 190°C; [α] <sub>D</sub> <sup>20</sup> = -10.29° (c = 26.23 mg/5 ml in methanol)
6	4-HOOC	2	H	CH <sub>3</sub>	H	3S-trans; mp. 175°C; [α] <sub>D</sub> <sup>20</sup> = -9.21° (c = 24.96 mg/5 ml in methanol)
7	2-HOOC	3	H	Cl	H	trans; mp. 148°C
8	4-HOOC	2	CH <sub>3</sub>	H	H	3S-trans; .HCl (1:1) .H <sub>2</sub> O (1:1); mp. >120°C; [α] <sub>D</sub> <sup>20</sup> = -13.52° (c = 24.04 mg/5 ml in methanol)
9	4-HOOC	3	H	Cl	H	trans; mp. 122°C
10	3-HOOC	3	H	Cl	H	trans
11	3-HOOC	3	CH <sub>3</sub>	H	H	3S-trans; mp. 120°C; [α] <sub>D</sub> <sup>20</sup> = -11.05° (c = 9.23 mg/2 ml in methanol)
12	3-HOOC	3	H	CH <sub>3</sub>	H	trans; mp. 122°C
13	4-HOOC	3	H	CH <sub>3</sub>	H	trans; mp. 118°C
14	4-HOOC	3	CH <sub>3</sub>	H	H	3S-trans; mp. 155°C; [α] <sub>D</sub> <sup>20</sup> = -8.03° (c = 12.20 mg/2 ml in methanol)
15	2-HOOC	3	CH <sub>3</sub>	H	H	3S-trans; .H <sub>2</sub> O (1:1); mp. 156°C; [α] <sub>D</sub> <sup>20</sup> = -13.34° (c = 11.54 mg/2 ml in methanol)
16	4-HOOC	3	Cl	Cl	H	trans; .HCl (1:1); mp. 256°C
17	4-HOOC	3	Br	Cl	H	trans; .HCl (1:1) .H <sub>2</sub> O (1:2); mp. 268°C
18	4-HOOC	3	Br	Cl	CH <sub>3</sub>	trans; .HCl (1:1) .H <sub>2</sub> O (2:3); mp. 136°C
19	4-HOOC	3	H	F	CH <sub>3</sub>	trans; mp. 187.7-198.6°C
20	4-HOOC	3	Cl	Cl	CH <sub>3</sub>	trans; mp. 200°C
21	4-HOOC	3	CH <sub>3</sub> O	H	CH <sub>3</sub>	trans; mp 179.4-189.2°C
22	4-HOOC	3	H	CH <sub>3</sub>	CH <sub>3</sub>	trans

Pharmacological examplesExample C.1 : "5HT<sub>4</sub> antagonism"

- h5-HT<sub>4b</sub>-HEK 293 clone 9 cells were cultured in 150 mm Petri dishes and washed twice with cold PBS. The cells were then scraped from the plates and suspended in
- 5 50 mM Tris-HCl buffer, pH 7.4 and harvested by centrifugation at 23,500 rpm for 10 minutes. The pellet was resuspended in 5 mM Tris-HCl, pH 7.4 and homogenized with an Ultra Turrax homogenizer. The membranes were collected by centrifugation at 30,000 rpm for 20 min, resuspended in 50 mM Tris-HCl pH 7.4 and stored at -80°C. For the experiment, assay mixtures (0.5 ml) contained 50 µl of the tritiated ligand
- 10 (5-HT<sub>4</sub> antagonist [<sup>3</sup>H]GR113808 0.1 nM) and 0.4 ml membrane preparation (15 µg protein/ml). 50 µl of 10% DMSO was added for total binding. 50 µl of 1 µM of (+)-trans-(1-butyl-3-hydroxy-4-piperidinyl)methyl 8-amino-7-chloro-2,3-dihydro-1,4-benzodioxin-5-carboxylate (a proprietary 5HT<sub>4</sub> agonist of Janssen Pharmaceutica) was added for determination of non-specific binding.
- 15 The [<sup>3</sup>H]GR113808 assay buffer was 50 mM HEPES-NaOH, pH 7.4. The mixtures were incubated for 30 min at 25°C. The incubation was terminated by filtration over a Unifilter 96 GF/B presoaked in 0.1% polyethylenimine, followed by six washing steps with 50 mM HEPES-NaOH, pH 7.4.
- Ligand concentration binding isotherms (rectangular hyperbola) were calculated by
- 20 nonlinear regression analysis and the pIC<sub>50</sub> data for all tested compounds are listed below in Table C.1.

Table C.1 : 5HT<sub>4</sub> antagonistic data

Co. No.	pIC <sub>50</sub>	Co. No.	pIC <sub>50</sub>
1	6.92	12	7.63
2	8.02	13	7.68
3	8.31	14	8.06
4	7.84	15	7.02
5	8.29	16	8.12
6	7.2	17	8.25
7	7.1	18	8.15
8	7.37	19	7.74
9	7.95	20	8.11
10	7.5	21	7.18
11	8.07	22	7.87

Example C.2 "Metabolic stability"

Sub-cellular tissue preparations were made according to Gorrod *et al.* (Xenobiotica 5: 453-462, 1975) by centrifugal separation after mechanical homogenization of tissue.

- 5 Liver tissue was rinsed in ice-cold 0.1 M Tris-HCl (pH 7.4) buffer to wash excess blood. Tissue was then blotted dry, weighed and chopped coarsely using surgical scissors. The tissue pieces were homogenized in 3 volumes of ice-cold 0.1 M phosphate buffer (pH 7.4).

- 10 Tissue homogenates were centrifuged at 9000 x g for 20 minutes at 4 °C. The resulting supernatant was stored at -80 °C and is designated 'S9'.

The S9 fraction can be further centrifuged at 100.000 x g for 60 minutes (4 °C). The resulting supernatant was carefully aspirated, aliquoted and designated 'cytosol'. The pellet was re-suspended in 0.1 M phosphate buffer (pH 7.4) in a final volume of 1 ml per 0.5 g original tissue weight and designated 'microsomes'.

- 15 All sub-cellular fractions were aliquoted, immediately frozen in liquid nitrogen and stored at -80 °C until use.

- For the samples to be tested, the incubation mixture contained PBS (0.1M), compound (5 µM), microsomes (1 mg/ml) and a NADPH-generating system (0.8 mM glucose-6-phosphate, 0.8 mM magnesium chloride and 0.8 Units of glucose-6-phosphate  
20 dehydrogenase). Control samples contained the same material but the microsomes were replaced by heat inactivated (10 minutes at 95 degrees Celsius) microsomes. Recovery of the compounds in the control samples was always 100%.

- The mixtures were preincubated for 5 minutes at 37 degrees Celsius. The reaction was started at time point zero (t = 0) by addition of 0.8 mM NADP and the samples were  
25 incubated for 60 minutes (t=60). The reaction was terminated by the addition of 2 volumes of DMSO. Then the samples were centrifuged for 10 minutes at 900 x g and the supernatants were stored at room temperature for no longer as 24 hours before analysis. All incubations were performed in duplo. Analysis of the supernatants was performed with LC-MS analysis. Elution of the samples was performed on a Xterra MS  
30 C18 (50 x 4.6 mm, 5 µm, Waters, US). An Alliance 2790 (Supplier: Waters, US) HPLC system was used. Elution was with buffer A (25 mM ammoniumacetate (pH 5.2) in H<sub>2</sub>O/acetonitrile (95/5)), solvent B being acetonitrile and solvent C methanol at a flow rate of 2.4 ml/min. The gradient employed was increasing the organic phase concentration from 0 % over 50 % B and 50 % C in 5 min up to 100 % B in 1 minute in  
35 a linear fashion and organic phase concentration was kept stationary for an additional 1.5 minutes. Total injection volume of the samples was 25 µl.

A Quatro triple quadrupole mass spectrometer fitted with an ESP source was used as detector. The source and the desolvation temperature were set at 120 and 350 °C respectively and nitrogen was used as nebuliser and drying gas. Data were acquired in positive scan mode (single ion reaction). Cone voltage was set at 10 V and the dwell time was 1 second.

Metabolic stability was expressed as % metabolism of the compound after 60 minutes (equation given as example) of incubation in the presence of active microsomes (E(act))

$$(\% \text{ metabolism} = 100 \% - \left( \frac{\text{Total Ion Current (TIC) of E(act) at } t = 60}{\text{TIC of E(act) at } t = 0} \right) \times 100).$$

Table C.2 : % metabolised compound after 60 minutes

Co. No.	% metabolized
2	6
3	16
5	15
7	14
9	7.5
10	13.5
11	1

Co. No.	% metabolized
12	0
14	0
15	6.5
16	3
17	6
18	20
22	6